A NEW METHOD OF SYNTHESIS OF STEREOSPECIFICALLY 7,7-DIFUNCTIONALIZED BICYCLO [2.2.1] HEPTENE DERIVATIVES

A.G. González, J. Darias and F. Díaz Instituto de Productos Naturales Orgánicos, CSIC, La Laguna; Instituto de Química Orgánica, Universidad de La Laguna, Teneriffe, Canary Islands, Spain

Abstract. - The cyclopropyl carbinol group of the adducts $\underline{7a}$ and $\underline{7b}$ undergoes anchimerically-assisted solvolytic ring opening, leading to a stabilized 7-norbornenyl cation which collapses, stereospecifically, to 7,7-difunctionalized bicyclic/2.2.1/heptene derivatives $\underline{8}$ and $\underline{9}$.

The current general methodology for the synthesis of 7-alkyl,7-hydroxy disubstituted bicyclo[2.2.1] heptene derivatives 1 involves the alkylation of a 7-norbornenone $\frac{2}{2}$ by an organometallic reagent^{1,2} (scheme I). Unfortunately, a direct synthesis of 2 by a Diels Alder reaction with cyclopentadienone 3 is not possible 3 and the synthesis of compounds of type 1 needs to be resolved by a circuitous method using a synthetic equivalent of 3 such as 4 or 5. However, the Diels Alder reaction of the dienone ketals 4 is limited to the use of the highly reactive dienophiles⁴ and, although the chlorinated diene⁻⁵ 5 reacts with a wide range of them,⁶ the attractiveness of this synthon is offset by the necessity to reductively dechlorinate the resulting adducts, despite recent efforts to improve the yields of these reactions.⁷

We were intrigued by the simplest approach to the bicyclic framework 1 as shown in the retrosynthetic analysis in scheme II. Since the difficulties associated with the synthesis of 5,5-disubstituted cyclopentadienes such as 6 are well-known⁸, attention was devoted to devising an alternative way equivalent to that shown in scheme II. The present investigation originated from the assumption that the solvolytic opening of the cyclopropane ring of compound 7 (scheme III) should be selectively directed by the driving force constituted by the generation of an anchimerically stabilized 7-norbornenyl cation.⁹ Nucleophilic capture of the intermediate should provide 7-disubstituted derivatives <u>8</u> and/or <u>9</u> stereospecifically.

To test the validity of the hypothesis, in view of the importance of compounds of type 1 in mechanistic and synthetic studies,¹⁰ particularly in the stereocontrolled construction of cyclopentane rings of biological interest by oxygen ring insertion 11 reactions in stereospecifically disubstituted norbornenones, we set out to prepare compounds of type 7.

The syntheses of compounds 7 a and 7 b were straightforward using a Diels Alder reaction, with quantitative yield, of the readily accessible¹² spirodiene 10 with methyl acrylate and methyl vinyl ketone respectively 13 followed by mesylation. 14 When the mesylates 7a and 7b were stirred on silica gel (Cl₂CH₂, 25°C) a fast reaction took place producing gratifying rearrangement products 8a, 9a and 8b, 9b in ratio 70:30 (85% yield from 10) and 72:28 (80% yield from 10) respectively. A dramatic increase in selectivity (99:1) was obtained when the rearrangement of 7a was performed with activated silica gel.

The stereochemistries of $\underline{8}$ and $\underline{9}$ were assigned on the basis of spectroscopical, mechanistic and chemical evidence. The single, crystalline compound 8b (mp 58-59°C) was converted to a mixture of endo and exo isomers¹⁵ from which the new epimer was chromatographically separated as

Scheme I



 R_1 , R_2 = H and/or EWG <u>9b</u>, R_1 =COCH₃, R_2 =H, X=OH

the less polar compound. Comparison of the ¹H-NMR spectra of both led us to assign the <u>exo</u> stereochemistry to the new isomer for the following reasons: 1) the higher field position of the methyl ketone signal in <u>8b</u> is attributable to shielding by the uplying endocyclic double bond, 2) the ring olefinic protons of the <u>endo</u> isomer appeared as two multiplets and in the <u>exo</u> isomer as a triplet because the acyl group is far enough removed from the olefinic centre to allow the olefinic protons to be equivalent.¹⁶ In the same way, <u>endo</u> stereochemistry was assigned to the acyl group of <u>8a</u> and to the mixture of adducts <u>7a</u> and <u>7b</u>, in which the hydroxymethyl group <u>syn</u> to the double bond was expected¹⁷ due to steric interaction of the diene-dienophile approach in the cycloaddition.

The retention of configuration in the interchange¹⁸ <u>8a</u> \neq <u>8c</u> agrees with the anchimeric participation of the endocyclic double bond and the exclusively <u>anti</u> attack of the nucleophile was chemically established by the formation of lactone <u>12</u> when the dianion of <u>8a</u> was treated with methyl disulfide.¹⁹ The ¹H-NMR spectrum of <u>8a</u> was essentially identical with that of <u>8c</u>, establishing the identities of the stereochemistries.

In the overall process described herein we can consider that the stable spirodiene <u>10</u> fulfils the function equivalent²⁰ to that of the difficult to achieve cyclopentadiene derivative <u>11</u> and opens an interesting way of access to stereospecifically disubstituted bicyclo[2.2.1] heptenes. Solvolysis of a secondary mesylate derivative and selective trapping of the intermediate by hydride, carbanion or other kinds of nucleophiles, should be an entry to a diversity of mono and dialkyl 7-substituted norbornenes. This in combination with a retro Diels Alder reaction would facilitate the synthesis of 5,5-dialkyl substituted cyclopentadienes, important synthons for which there is no satisfactory methodology²¹ at present. Efforts in these directions are under way in these laboratories.



ACKNOWLEDGEMENT. Our work was generously supported by the Comisión Asesora de Investigación Científica y Técnica through the research grants nos. 22106-09 and 613/106.

REFERENCES

- J.A. Berson, T. Miyashi and G. Jones II, J. Amer. Chem. Soc., 96, 3468 (1974). J.A. Berson, M. Jones Jr., <u>ibid</u>, 86, 5017, 5019 (1964). J.A. Berson and E.D. Walsh Jr., <u>ibid</u>, 90, 4732 (1968). R.K. Bly and R.S. Bly, <u>J. Org. Chem</u>., <u>28</u>, 3165 (1963).
- 2 For other procedures see: (a) L.A. Paquette, L.W. Hertel, R. Gleiter, M.C. Bohm, M.A. Beno and G.C. Christoph, <u>J. Amer. Chem. Soc.</u>, <u>103</u>, 7106 (1981). (b) L. Skattebøl, <u>Tetrahedron</u>, <u>23</u>, 1107 (1967).
- 3 (a) C.D. De Puy, M. Isaks, K.L. Eilers and G.F. Morris, J. Org. Chem., 29, 3503, 3508 (1964).
 (b) P.E. Eaton and R.H. Hudson, J. Amer. Chem. Soc., 87, 2769 (1965) and references cited therein.
- 4 Ref. (3b). T. Baasov and B. Fuchs, <u>Tetrahedron Letters</u>, 1373 (1982). H. Tanida, T. Tsuji

and R. Irie, <u>J. Amer. Chem. Soc.</u>, <u>89</u>, 1953 (1967). J.S. Haywood-Farmer and R.E. Pincok, <u>ibid</u>, <u>91</u>, 3020 (1969). R. Muneyuki, T. Yano and H. Tanida, <u>ibid</u>, <u>91</u>, 2408 (1969). J.C. Barborak and R. Pettit, <u>ibid</u>, <u>89</u>, 3080 (1967). G.D. Sargent and M.A. Herkenham, <u>ibid</u>, <u>94</u>, 2892 (1972).

- 5 J.S. Newcomer and E.T. McBee, <u>J. Amer. Chem. Soc.</u>, 71, 946 (1949).
- 6 M.E. Jung and J.F. Hudspeth, J. Amer. Chem. Soc., 99, 5508 (1977) and references cited therein. <u>ibid</u>, <u>100</u>, 4309 (1978). L.A. Paquette, R.E. Moerck, B. Harirchian and P.D. Magnus, J. Amer. Chem. Soc., <u>100</u>, 1597 (1978). M.E. Jung and L.A. Light, <u>J. Org. Chem.</u>, <u>47</u>, 1084 (1982). M.E. Jung and C.D. Radcliffe, <u>Tetrahedron Letters</u>, 4397 (1980).
- 7 M.N. Paddon-Row and B.V. Lap, <u>J. Org. Chem.</u>, <u>44</u>, 4979 (1979). P.E. Eaton, R.S. Sidhu, G.E. Langford, D.A. Cullison and L. Cornel, <u>Tetrahedron</u>, <u>37</u>, 4479 (1981).
- 8 For reviews see: C.W. Spangler, <u>Chem. Rev.</u>, <u>76</u>, 187 (1976). R.F. Childs, <u>Tetrahedron</u>, <u>38</u>, 567 (1982).
- 9 S. Wistein, M. Shatavsky, C. Norton and R.B. Woodward, J. Amer. Chem. Soc., 77, 4183 (1955).
- 10 Ref. (1). P.A. Grieco, Y. Yokoyama, G.P. Withers, F.J. Okuniewicz and C.-L. J. Wang, <u>J. Org. Chem.</u>, <u>43</u>, 4178 (1978). C.-L. J. Wang, P.A. Grieco and F.J. Okuniewicz, <u>J. Chem. Soc.</u>, <u>Chem. Commun.</u>, <u>468</u> (1976). P.A. Grieco, W. Owens, C.-L. J. Wang, S. Williams and W.J. Schillinger, <u>J. Med. Chem</u>., <u>23</u>, 1072 (1980).
- 11 G.R. Krow, Tetrahedron, 37, 2697 (1981).
- 12 K. Bangert and V. Boekelheide, <u>Tetrahedron Letters</u>, 1119 (1963). H. Schaltegger, <u>Helv. Chim</u>. <u>Acta</u>, <u>45</u>, 1368 (1962).
- 13 Spirodiene 10 was stirred with two equiv. of methyl acrylate or methyl vinyl ketone overnight.
- 14 R.K. Crossland and K.L. Servis, J. Org. Chem., 35, 3195 (1970).
- 15 Quenching enolate equilibration generated from HNa in THF at 25°C.
- 16 P. Laszlo and P.R. Schleyer, J. Amer. Chem. Soc., <u>85</u>, 2709 (1963). J.G. Dinwiddie and S.P. McManus, <u>J. Org. Chem.</u>, <u>30</u>, 766 (1965). F.A.L. Anet, H.H. Lee and J.L. Sudmeier, <u>J. Amer. Chem. Soc.</u>, <u>89</u>, 4431 (1967).
- 17 E.J. Corey, P.R. Shiner, R.P. Volante and C.R. Cyr, Tetrahedron Letters, 1161 (1975).
- 18 Compound <u>8c</u> by stirring on silica gel was quantitatively converted to <u>8a</u>, from which chloride <u>8c</u> was regenerated by treatment with a methylene chloride solution of hydrogen chloride.
- 19 B.M. Trost and Y. Tamaru, J. Amer. Chem. Soc., 99, 3101 (1977).
- 20 The retro Diels Alder reaction of compounds <u>8a</u> and <u>8b</u> to produce <u>11</u> was clearly shown by the MS spectra giving the corresponding fragment at m/z = 122.
- 21 R.W. Holder, J.P. Daub, E.W. Baker, R.H. Gilbert III and N.A. Graf, J. Org. Chem., 47, 1445 (1982). J.W. Wilt and S.Z. Ahmed, <u>ibid</u>, 44, 4000 (1979). Ch.W. Jefford, T.W. Wallace, N.-T. H. Can and C.G. Rimbault, <u>ibid</u>, 44, 689 (1979) and references cited therein.

(Received in UK 6 March 1984)